



Lewatit® MDS TP 260 is a weakly acidic, macroporous cation exchange resin with chelating amino methyl phosphonic acid groups designed for the selective removal of heavy metal and alkaline earth cations. It is a small bead variant of **Lewatit® MonoPlus TP 260** with excellent kinetic performance. Divalent cations are removed from neutralized waters in following order (decreasing affinity):

Uranium (UO₂²⁺) > Lead > Copper > Zinc > Nickel > Cadmium > Cobalt >> Calcium > Magnesium > Strontium > Barium >>> Sodium.

The monodisperse, uniform sized beads of **Lewatit® MDS TP 260** are mechanically and osmotically more stable than ion exchange resin beds with a heterodisperse bead size distribution. Additionally they offer superior kinetic behavior which leads to faster uptake of cations and a better utilization of capacity. Due to its modified polymer structure and substitution grade it is suitable for the following applications:

- fine polishing of brine fed to chloralkali membrane cells, e.g. by removal of Ca²⁺, Mg²⁺, Ba²⁺, Sr²⁺; in the absence of Fe³⁺ ions
- antimony (Sb) and bismuth (Bi) removal from copper containing electrolyte
- · uranium (U) removal from crude phosphoric acid
- titanium (Ti) removal from recycled battery acid
- aluminum (AI) removal from urea solutions
- lead (Pb) and strontium (Sr) removal from BF₄ containing waste water out of PCB production
- removal of iron(II), nickel and zinc from 5 % gluconate containing liquid metal working solutions

Smaller bead size requires adaption of the linear velocity and duration for backwash, downflow conditioning and suspended solid load control.

Note:

In secondary brine purification of brine fed to chloralkali electrolyzer membranes Lewatit® MDS TP 260 offers following advantages compared to the standard Lewatit® MonoPlus TP 260 and Lewatit® TP 260:

- up to 100 % higher operating capacity without requiring additional regenerant chemicals
- · lower leakage levels
- · improved Strontium and Barium removal efficiency

The special properties of this product can only be fully utilized if the technology and process used correspond to the current state-of-the-art. Further advice in this matter can be obtained from Lanxess, Business Unit Liquid Purification Technologies.

This document contains important information and must be read in its entirety.





Common Description

Delivery form	Na⁺
Functional group	Aminomethylphosphonic
	acid
Matrix	Styrenic
Structure	Macroporous
Appearance	Beige, opaque

Specified Data

Uniformity coefficient		max.	1.15
Mean bead size	d50	mm	0.40 (+/- 0.04)
Total capacity (H ⁺ form)		min. eq/L	3.0

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Typical Physical and Chemical Properties

Bulk density for shipment (+/- 5%)	g/L	740
Density	approx. g/mL	1.21
Water retention (delivery form)	approx. weight %	63
Volume change (Na ⁺ - H ⁺)	max. approx. %	-35
Stability pH range		0-14
Stability temperature range	°C	1-80
Storage temperature range	°C	-20 - +40

Operation

Operating temperature		max. °C	80
Operating pH range	during exhaustion		1-12
Bed depth for single column		min. mm	1000
Back wash bed expansion per m/h (20°C)		%	15
Specific pressure loss kPa*h/m² (15°C)		kPa*h/m² (15°C)	3.0
Max. pressure loss during operation		kPa	250
Specific flow rate		max. BV/h	5-25
Freeboard	during backwash	min. vol. %	100

Regeneration

HCI regeneration	concentration	approx. wt. %	4-10
HCI regeneration	quantity co-current	min. g/L resin	150
Regeneration contact		min. minutes	20
time			
Slow rinse at		min. BV	4
regeneration flow rate			

Conditioning

NaOH conditioning	concentration	approx. wt. %	4
NaOH conditioning, di-	quantity	min. g/l resin	80-96
Na⁺			
Conditioning contact time		min. minutes	20
Slow rinse	at conditioning flow rate	min. BV	4

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Additional Information & Regulations

Safety precautions

Strong oxidants, e.g. nitric acid, can cause violent reactions if they come into contact with ion exchange resins

Toxicity

The safety data sheet must be observed. It contains additional data on product description, transport, storage, handling, safety and ecology.

Disposal

In the European Community Ion exchange resins have to be disposed, according to the European waste nomenclature which can be accessed on the internet-site of the European Union.

Storage

It is recommended to store ion exchange resins at temperatures above the freezing point of water under roof in dry conditions without exposure to direct sunlight. If resin should become frozen, it should not be mechanically handled and left to thaw out gradually at ambient temperature. It must be completely thawed before handling or use. No attempt should be made to accelerate the thawing process.

Packaging

The experience has shown that the packaging stability for reliable resin containment is limited to 24 months under the storage conditions described above. It is therefore recommended to use the product within this time frame; otherwise the packaging condition should be checked regularly.



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