

Product Information



DOWEX 88

Ion Exchange Resin for Sweetener Applications

Product	Туре	Matrix	Functional group
DOWEX™ 88	Strong acid cation	Styrene-DVB, macroporous	Sulfonate

Typical Physical and Chemical Properties

lonic form as produced		Na+	
Total exchange capacity, min.	eq/L	1.8	
Water content	%	42 - 48	
Bead size distribution			
Range	μm	300 - 1,200	
> 1,200 µm 16 mesh	%	< 5	
< 420 µm 40 mesh	%	< 5	
Total swelling (Na ⁺ → H ⁺)	%	5	
Whole uncracked beads, min.	%	95	
Particle density, approx.	g/mL	1.2	
Shipping weight, approx.	g/L	800	
2	lbs/ft³	50	

Recommended Operating Conditions

 Maximum operating temperature (H⁺ form) 	150°C (300°F)
• pH range	0 - 14
Bed depth, min.	91 cm (3 ft)
 Flow rates: Service Backwash Regeneration time Displacement rinse Fast rinse (if applicable) 	2 - 4 bed volumes/hour See Figure 1 30 - 45 min. 30 - 45 min. 2 - 10 bed volumes/hour
Total rinse requirement	3 - 6 bed volumes
 Regenerant: Concentration Level, 100% basis[†] 	5 - 7% HCl 6 - 7 lbs/ft³

96 - 112 kg/m³ 93°C (200°F)

Form No. 177-01684-0705

Temperature, max.

[†] Regeneration level may be lower for counter-current regeneration systems.

Typical Properties and Applications

DOWEX 88 resin is a macroporous strong acid cation resin for use in deashing, softening and demineralization. This macroporous matrix provides excellent mechanical strength and good operating capacity.

Packaging

25 liter bags, 5 cubic feet fiber drums or 1 cubic meter super sacks.

Figure 1. Backwash Expansion Data

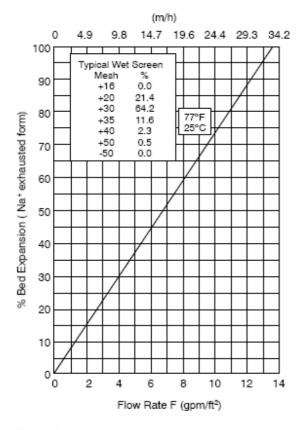
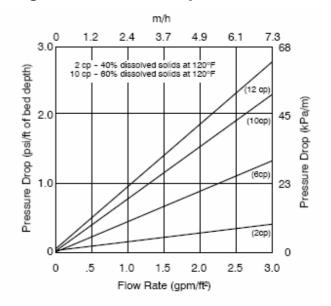


Figure 2. Pressure Drop Data



For other temperatures use:

 $P_T = P_{20^{\circ}C} / (0.026 \, T_{\circ C} + 0.48)$, where $P \equiv bar/m$ $P_T = P_{68^{\circ}F} / (0.014 \, T_{\circ F} + 0.05)$, where $P \equiv psi/ft$

For other temperatures use:

 $F_T = F_{77^{\circ}F} [1 + 0.008 (T_{\circ}F - 77)], \text{ where } F \equiv \text{gpm/ft}^2$ $F_T = F_{25^{\circ}C} [1 + 0.008 (1.8T_{\circ}C - 45)], \text{ where } F \equiv \text{m/h}$

Warning: Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

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