

**Product Data Sheet** 



## AMBERSEP<sup>™</sup> 400 SO<sub>4</sub> Ion Exchange Resin

Industrial-grade Strong Base Anion Exchanger

Description	AMBERSEP <sup>™</sup> 400 SO <sub>4</sub> lon Exchange Resin is a gel, Type I strong basic anion exchange resin with outstanding performance for uranium recovery. Its excellent selectivity for the uranyl sulfate ion over other anions, high operating capacity, excellent mechanical and physical stability, and its resistance to fouling make it the resin of choice. AMBERSEP <sup>™</sup> 400 SO <sub>4</sub> is well-suited for the recovery of uranium from sulfuric acid leach systems using fixed beds, <i>in situ</i> leaching, fluidized beds, or Resin In Pulp (RIP) applications. AMBERSEP <sup>™</sup> 400 SO <sub>4</sub> is supplied in the sulfate form in order to minimize the presence of chloride upon start-up.	
	If used in sulfuric acid leach systems, no preconditioning of this resin is required and the resin can be used as supplied.	
Applications	<ul> <li>Uranium extraction from sulfuric acid leach systems</li> <li>Fixed beds</li> <li><i>In situ</i> leaching</li> <li>Fluidized beds</li> <li>Resin In Pulp (RIP) systems</li> </ul>	
<b>Typical Properties</b>	Physical Properties	
	Copolymer	Styrene-divinylbenzene
	Matrix	Gel
	Туре	Strong base anion, Type I
	Functional Group	Trimethylammonium
	Physical Form	Amber, translucent, spherical beads
	Chemical Properties	
	Ionic Form as Shipped	SO4 <sup>2-</sup>
	Total Exchange Capacity	≥ 1.40 eq/L (Cl⁻ form)
	Water Retention Capacity	40 – 47% (Cl⁻ form)
	Particle Size §	
	Particle Diameter	600 – 750 μm
	Uniformity Coefficient	≤ 1.60
	< 500 μm	≤ 1.0%
	> 1180 µm	≤ 5.0%
	Density	
	Shipping Weight	730 g/L

§ For additional particle size information, please refer to the <u>Particle Size Distribution Cross Reference Chart</u> (Form No. 177-01775).

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	<ul> <li>Please be aware of the following:</li> <li>WARNING: Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to</li> </ul>

under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.



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