

Product Data Sheet



AmberLite™ FPC14 Na Ion Exchange Resin

Food-grade, Gel, Strong Acid Cation Exchange Resin

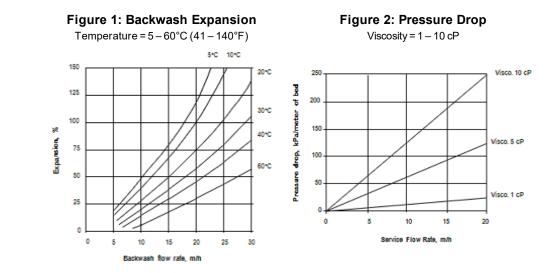
Description	AmberLite [™] FPC14 Na Ion Exchange Resin is a gel, strong acid cation exchanger that has been developed with a special, solvent-free manufacturing process. With its excellent physical, chemical, and thermal stability, and also good ion exchange kinetics and high exchange capacity, AmberLite [™] FPC14 Na is useful for pharmaceutical and food applications such as decalcification of saccharose thin juice or amino acid recovery.	
Applications	Saccharose thin juice decalcifAmino acid recovery	ïcation
Typical Properties	Physical Properties	_
	Copolymer	Styrene-divinylbenzene
	Matrix	Gel
	Туре	Strong acid cation
	Functional Group	Sulfonic acid
	Physical Form	Amber, translucent, spherical beads
	Chemical Properties	·
	Ionic Form as Shipped	Na ⁺
	Total Exchange Capacity	≥ 1.90 eq/L
	Water Retention Capacity	41-49%
	Particle Size [§]	
	Particle Diameter	600 – 800 μm
	< 300 µm	≤2.0%
	Density	
	Shipping Weight	808 g/L
	[§] For additional particle size information, p (Form No. 45-D00954-en).	lease refer to the <u>Particle Size Distribution Cross Reference Chart</u>
Suggested	Maximum Operating Temperature	120°C (248°F)
Operating	Flowrates	
Conditions	Service	5–20 BV*/h
	Regeneration	1 – 3 BV/h
	Backwash	See Figure 1
	Slow Rinse	Regeneration flowrate for 2 BV
	Fast Rinse (if applicable)	Service flowrate for 2-4 BV
	Contact Time	
	Regeneration	≥ 30 – 45 minutes
	Regenerant	NaCl
	Concentration	10%
	Level, 100% basis	$60 - 250 \text{ kg/m}^3$
		(3.8–15.6 lb/ft ³)

* 1 BV (Bed Volume) = 1 m³ solution per m³ resin or 7.5 gal solution per ft³ resin

Hydraulic Characteristics

Bed expansion of AmberLite[™] FPC14 Na Ion Exchange Resin as a function of backwash flowrate and temperature is shown in Figure 1.

Pressure drop data for AmberLite[™] FPC14 Na as a function of service flowrate and viscosity is shown in Figure 2. These pressure drop expectations are valid at the start of the service run with clean feed and a well-classified bed.



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Please be aware of the following:

• **WARNING:** Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.



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