

Product Data Sheet



## AmberLite<sup>™</sup> MB20 H/OH Ion Exchange Resin

Mixture of Gaussian, Gel, Strong Acid Cation and Strong Base Anion Exchange Resins for Industrial Demineralization Applications

| Description  | AmberLite <sup>™</sup> MB20 H/OH Ion Exchange Resin is an equilibrated, homogeneous mixture of a dark strong acid cation and a clear strong base anion exchange resins. It is fully regenerated, ready-to-use, pre-mixed resin developed for the production of high-purity water in working and mixed bed polishing applications. The pre-mixed resin also allows for faster initial rinse-up prior to service, which minimizes rinse wastewater volume.   |
|--------------|--|
|              | AmberLite <sup>TM</sup> MB20 H/OH is most commonly used in service deionization for a full demineralization of water when removal of silica and $CO_2$ is required. In most of the applications, the conductivity of the treated water is much lower than 0.1 µS/cm and the pH is neutral. If necessary, the resin can be regenerated after exhaustion. Both components must be separated by backwashing and regenerated separately.   |
|              | AmberLite <sup>™</sup> MB20 H/OH is the reference mixed bed for service deionization. The resin mixture is prepared from high-quality components and the proprietary manufacturing process ensures consistency from batch to batch. This enables the resin to perform in a highly stable manner delivering high-quality treated water consistently in both working and polishing mixed beds. The consistency in quality combined with visible separation of cation and anion resins prior to regeneration make AmberLite <sup>™</sup> MB20 H/OH a trusted choice for mixed bed pool systems. |
| Applications | <ul> <li>Service deionization</li> <li>Working mixed bed on tap water in small installations</li> <li>Mixed bed polishing on RO or demineralized water</li> </ul>  |

- System Designs
  Externally-regenerated mixed beds
  Non-regenerated mixed beds
- HistoricalAmberLite™ MB20 H/OH Ion Exchange Resin has previously been sold asReferenceAmberLite™ MB20 Ion Exchange Resin.

#### **Typical Properties**

|                       | Cation Resin             | Anion Resin               |
|-----------------------|--------------------------|---------------------------|
| Physical Properties   |                          |                           |
| Copolymer             | Styrene-divinylbenzene   | Styrene-divinylbenzene    |
| Matrix                | Gel                      | Gel                       |
| Туре                  | Strong acid cation       | Strong base anion, Type I |
| Functional Group      | Sulfonic acid            | Trimethylammonium         |
| Physical Form         | Dark amber, translucent, | Clear amber, translucent, |
|                       | spherical beads          | spherical beads           |
| Volume Ratio          | 38-44%                   | 62-56%                    |
| Chemical Properties   |                          |                           |
| Ionic Form as Shipped | H⁺                       | OH⁻                       |
| Particle Size §       |                          |                           |
| < 300 µm              | ≤3.0%                    |                           |
| Density               |                          |                           |
| Shipping Weight       | 710 g/L                  |                           |

<sup>§</sup> For additional particle size information, please refer to the <u>Particle Size Distribution Cross Reference Chart</u> (Form No. 45-D00954-en).

# Operating Capacity

The operating capacity of AmberLite<sup>™</sup> MB20 H/OH Ion Exchange Resin can be estimated using the following formula, which gives an approximate determination of volume of water that can be treated:

$$BV = \frac{500}{TDS (meq/L)} \qquad \text{or} \qquad \frac{gal}{ft^3} = \frac{187000}{TDS (as ppm CaCO_3)}$$

where BV (Bed Volume) is the number of liters of a feedwater containing a TDS (Total Dissolved Solids) given in meq/L that can be demineralized with one liter of the resin mixture when run to exhaustion (or US gallons per cubic foot of the resin with TDS as ppm CaCO<sub>3</sub>).

#### **Treated Water Quality**

AmberLite<sup>TM</sup> MB20 H/OH Ion Exchange Resin provides a high-quality demineralized water with a conductivity < 0.1  $\mu$ S/cm and neutral pH that will satisfy most of the cartridge and laboratory applications.

| Temperature Range (H <sup>+</sup> /OH <sup>-</sup> form) <sup>‡</sup> | 5-60°C (41-140°F) |
|---|-------------------|
| pH Range  | 0 – 14            |

 $^{\ddagger}$  Operating mixed beds at elevated temperatures, for example above  $60 - 70^{\circ}C (140 - 158^{\circ}F)$ , may impact the purity of the loop and resin life. Contact our technical representative for details.

For additional information regarding recommended minimum bed depth, operating conditions, and regeneration conditions for <u>mixed beds</u> (Form No. 45-D01127-en) or <u>separate beds</u> (Form No. 45-D01131-en) in water treatment, please refer to our Tech Facts.

### Suggested Operating Conditions

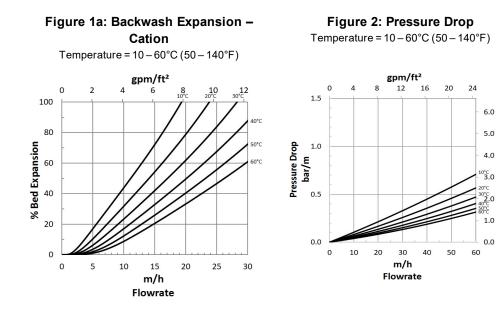
Product

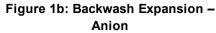
Performance

## Hydraulic Characteristics

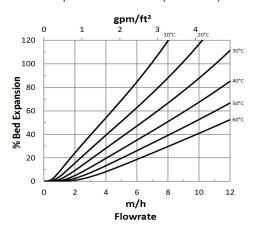
Estimated bed expansion of the cation component (Figure 1a) and of the anion component (Figure 1b) of AmberLite<sup>™</sup> MB20 H/OH Ion Exchange Resin as a function of backwash flowrate and temperature are shown.

Estimated pressure drop for AmberLite<sup>™</sup> MB20 H/OH as a function of service flowrate and temperature is shown in Figure 2. These pressure drop expectations are valid at the start of the service run with clean water and a well-classified bed.





Temperature =  $10 - 60^{\circ}C(50 - 140^{\circ}F)$ 



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psi/ft

#### **Customer Notice**

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Please be aware of the following:

• **WARNING:** Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.



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