



# DOWEX UPCORE Mono A-500

**A Uniform Particle Size, Strong Base Anion Exchange Resin Specifically Designed for the UPCORE System**

Product	Type	Matrix	Functional group
DOWEX* UPCORE* Mono A-500	Type 1 strong base	Styrene-DVB, gel	Quaternary amine

Guaranteed Sales Specifications		Cl <sup>-</sup> form
Total exchange capacity, min.	eq/l	1.3
	kgr/ft <sup>3</sup> as CaCO <sub>3</sub>	28.4
Water content	%	50 - 58
Bead size distribution <sup>†</sup>		
Mean particle size	µm	575 ± 50
Uniformity coefficient, max.		1.1
>850µ, max.	%	5
<300µ, max.	%	0.5
Whole uncracked beads, min.	%	95

Typical Physical and Chemical Properties		Cl <sup>-</sup> form
Total swelling (Cl <sup>-</sup> -OH <sup>-</sup> )	%	20
Particle density	g/ml	1.08
Shipping weight	g/l	670
	lbs/ft <sup>3</sup>	42

Recommended Operating Conditions	
Maximum operating temperature:	
OH <sup>-</sup> form	60°C (140°F)
Cl <sup>-</sup> form	100°C (212°F)
pH range	0-14
Bed depth, min.	1200 mm (4 ft)
Pressure drop, design max.	1.5 bar (22 psi)
Pressure drop, max.	2.5 bar (37 psi)
Flow rates:	
Service/fast rinse	5-60 m/h (2-24 gpm/ft <sup>2</sup> )
Regeneration/displacement rinse	4-10 m/h (1.6-4 gpm/ft <sup>2</sup> )
Total rinse requirement	2-4 Bed volumes
Regenerant	2-5% NaOH

<sup>†</sup>For additional particle size information, please refer to the Particle Size Distribution Cross Reference Chart (Form No 177-01775/CH 171-476-E).

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### Typical properties and applications:

DOWEX\* UPCORE\* Mono A-500 strong base anion resin is a uniform particle size, gellular, type 1 anion resin designed for use in the UPCORE packed bed counter-current regeneration system.

The absence of large beads in DOWEX UPCORE Mono A-500 resin

results in high operating capacity and good resistance to silica fouling.

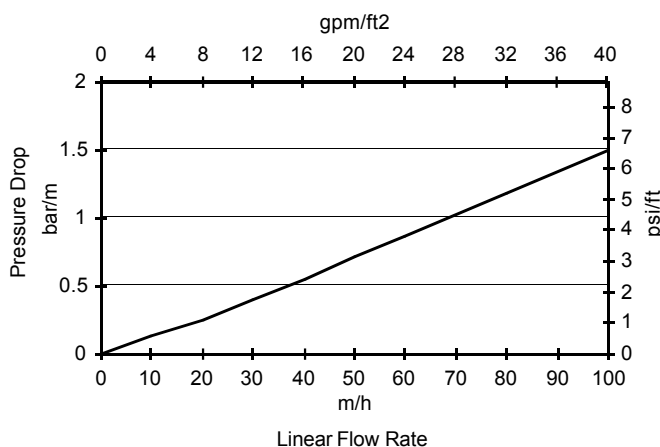
DOWEX UPCORE Mono A-500 resin has an excellent resistance to mechanical and osmotic stress which helps minimize resin attrition.

### Packaging

25 liter bags or 5 cubic feet fiber drums.

**Figure 1. Pressure Drop Data**

Temperature = 20° C (68° F)



### For other temperatures use:

$$P_T = P_{20^\circ\text{C}} / (0.026 T_{\text{C}} + 0.48), \text{ where } P \equiv \text{bar/m}$$

$$P_T = P_{68^\circ\text{F}} / (0.014 T_{\text{F}} + 0.05), \text{ where } P \equiv \text{psi/ft}$$

**Warning:** Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

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