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8-12% NaCl

# **DOWEX™ MARATHON™ MSC**

A Uniform Particle Size, High Capacity Macroporous Cation Exchange Resin for Industrial Softening and Water Demineralization Applications

Product	Туре	Matrix	Functional group
DOWEX™ MARATHON™ MSC	Strong acid cation	Styrene-DVB, macropore	ous Sulfonic acid
Guaranteed Sales Specification	ns	Na⁺ form	H+ form
Total exchange capacity, min.	eq/L	1.7	1.6
	kgr/ft³ as Ca	CO₃ 37.1	35.0
Water content	%	44 - 50	50 - 56
Uniformity coefficient, max.		1.1	1.1
Typical Physical and Chemical	Properties	Na⁺ form	H+ form
Mean particle size	μm	550 ± 50	575 ± 50
Whole beads	%	95 - 100	95 - 100
Total swelling (Na <sup>+</sup> → H <sup>+</sup> )	%	4	4
Particle density	g/mL	1.28	1.20
Shipping weight	g/L	800	760
	lbs/ft³	50	47
Recommended Operating Conditions	<ul> <li>Maximum operating tempera</li> <li>pH range</li> </ul>	ture	150°C (300°F) 0 - 14
	<ul> <li>Bed depth, min.</li> <li>Flow rates:         Service/fast rinse         Backwash         Co-current regeneration/disp</li> </ul>		800 mm (2.6 ft)  5-50 m/h (2-20 gpm/ft²) see Figure 1 1-10 m/h (0.4-4 gpm /ft²)
	Counter-current regeneration     Total rinse requirement	n/displacement rinse	5-20 m/h (2-8 gpm /ft²) 3 - 6 Bed volumes
	<ul> <li>Regenerant</li> </ul>		1-10% H <sub>2</sub> SO <sub>4</sub> , 4-8% HCl or

<sup>†</sup> For additional particle size information, please refer to Particle Size Distribution Cross Reference Chart (Form No. 177-01775).

# Typical Properties and Applications

DOWEX™ MARATHON™ MSC strong acid cation resin is a highly cross-linked resin with high porosity giving excellent osmotic shock resistance and chemical and thermal stability.

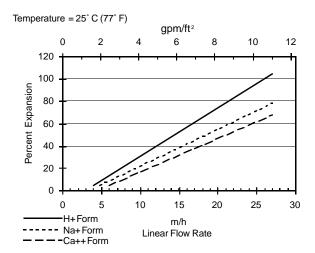
The resin has a variety of uses, such as:

- Hot process softening
- Demineralization
- Adsorbent
- Processes with oxidizing conditions
- Recovery of metals from plating baths

## **Packaging**

25 liter bags or 5 cubic feet fiber drums

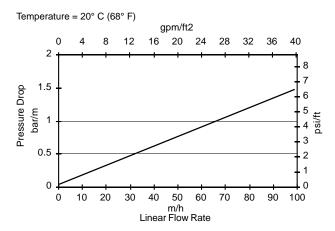
# Figure 1. Backwash Expansion Data



#### For other temperatures use:

 $F_T = F_{77^{\circ}F} [1 + 0.008 (T_{\circ F} - 77)], \text{ where } F = \text{ gpm/ft}^2$  $F_T = F_{25^{\circ}C} [1 + 0.008 (1.8T_{\circ C} - 45)], \text{ where } F = \text{ m/h}$ 

### Figure 2. Pressure Drop Data



#### For other temperatures use:

 $P_T = P_{20^{\circ}C} / (0.026 \ T_{^{\circ}C} + 0.48)$ , where P = bar/m $P_T = P_{68^{\circ}F} / (0.014 \ T_{^{\circ}F} + 0.05)$ , where P = psi/ft

Warning: Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

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