DOWEX HCR-S/S

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A High Capacity Cation Exchange Resin for Domestic Applications

Product	Туре	Matrix	Functional group
DOWEX* HCR-S/S	Strong acid cation	Styrene-DVB gel	Sulfonic acid
Guaranteed Sales Specif	ications		Na ⁺ form
Total exchange capacity, m	in.	eq/l	1.9
		kgr/ft³ as CaCO₃	41.5
Bead size distribution range	9 [†]		
0.3 - 1.2 mm, min.		%	90
<0.3 mm, max.		%	1
Whole uncracked beads, m	in.	%	90
Color throw, as packaged, r	nax.	APHA	20
Acidity range		рН	7.0 - 9.5

Typical Physical and Chemical Properties		Na ⁺ form	
Water content	%	48 - 52	
Total swelling (Ca ⁺ \rightarrow Na ⁺)	%	5	
Particle density	g/ml	1.30	
Shipping weight	g/l Ibs/ft ³	800 50	

Recommended Operating Conditions			
Maximum operating temperature	120°C (250°F)		
pH range	0-14		
Bed depth, min.	800 mm (2.6 ft)		
Flow rates: Service/fast rinse Backwash Co-current regeneration/displacement rinse	5-50 m/h (2-20 gpm/ft²) See Figure 1 1-10 m/h (0.4-4 gpm/ft²)		
Total rinse requirement	3-6 Bed volumes		
Regenerant	8-12% NaCl		

[†]For additional particle size information, please refer to the Particle Size Distribution Cross Reference Chart (Form No. 177-01775/CH 171-476-E).

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DOWEX Ion Exchange Resins

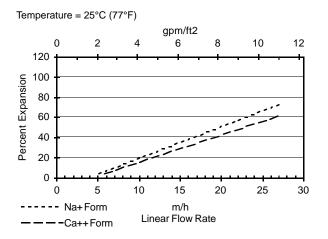
Typical properties and applications:

Packaging

25 liter bags or 1 cubic foot bags.

DOWEX HCR-S/S cation exchange resin is a high capacity resin with excellent kinetics and good physical, chemical, and thermal stability. DOWEX HCR-S/S is used for domestic applications in the co-current mode of regeneration. For counter-current regeneration, DOWEX HCR-S/S CR is available.

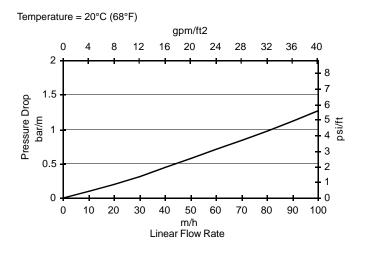
Figure 1. Backwash Expansion Data



For other temperatures use:

$$\begin{split} F_T &= F_{77^\circ F} \; [1+\; 0.008 \; (T_{^\circ F} \; -77)], \; \text{where} \; F \equiv gpm/ft^2 \\ F_T &= F_{25^\circ C} \; [1+\; 0.008 \; (1.8T_{^\circ C} \; -45)], \; \text{where} \; F \equiv m/h \end{split}$$

Figure 2. Pressure Drop Data



For other temperatures use:

 $P_T = P_{20^{\circ}C} / (0.026 T_{\circ C} + 0.48)$, where $P \equiv bar/m$ $P_T = P_{68^{\circ}F} / (0.014 T_{\circ F} + 0.05)$, where $P \equiv psi/ft$

Warning: Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

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