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Functional group

640

40



Product

DOWEX™ 66

Type

Ion Exchange Resin for Sweetener Applications

DOWEX™ 66	Weak base anion	Styrene-DVB, macroporous	Tertiary amine
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Typical Physical and Chemica	al Properties		FB (free base) form
Total exchange capacity, min.	eq/L		1.6
Weak base capacity, min.	eq/L		1.35
Water content	%		40 - 46
Bead size distribution			
Range	μm		300 - 1,200
> 1,200 μm (16 mesh)	%		< 2
< 300 μm (50 mesh)	%		< 3
Total swelling (FB → HCl-)	%		20
Particle density, approx.	g/mL		1.04

g/L

lbs/ft3

Matrix

Recommended **Operating Conditions**

Shipping weight**, approx.

 Maximum operating temperature (OH·) pH range 	60°C (140°F) 0 - 7
Bed depth, min.: Flow rates: Service Backwash Regeneration time Displacement rinse Fact rings (if applicable)	910 mm (3 ft) 2 - 4 bed volumes/hour See Figure 1 30 - 45 min. 30 - 45 min.
Fast rinse (if applicable) Total rinse requirement	2 - 10 bed volumes/hour 4 - 6 bed volumes

Regenerants	NaOH [†]	Na ₂ CO ₃	NH ₄ OH	
Concentration (%)	4	5	5	
Level, 100% basis ^{††}				
lbs/ft ³	5 - 6	7 - 8	5 - 6	
kg/m³	80 - 96	112 - 128	80 - 96	
Temperature, max.	60°C (140°F)	60°C (140°F)	60°C (140°F)	

^{††} Regeneration level may be lower for counter-current regeneration systems.
*****As per the backwashed and settled density of the resin, determined by ASTM D-2187

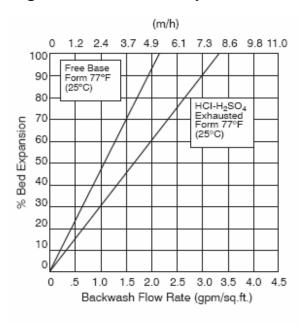
Typical Properties and Applications

DOWEX™ 66 resin is a macroporous weak base anion resin for use in the deashing (demineralization) applications. The unique resin matrix provides high operating capacity and excellent mechanical strength.

Packaging

25 liter bags, 5 cubic foot fiber drums or 1 cubic meter super sacks

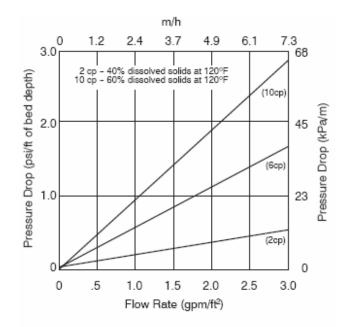
Figure 1. Backwash Expansion Data



For other temperatures use:

 $F_T = F_{77^{\circ}F} [1 + 0.008 (T_{\circ}F - 77)], \text{ where } F \equiv gpm/ft^2$ $F_T = F_{25^{\circ}C} [1 + 0.008 (1.8T_{\circ}C - 45)], \text{ where } F \equiv m/h$

Figure 2. Pressure Drop Data



For other temperatures use:

 $P_T = P_{20^{\circ}C} / (0.026 \, T_{^{\circ}C} + 0.48)$, where $P \equiv bar/m$ $P_T = P_{68^{\circ}F} / (0.014 \, T_{^{\circ}F} + 0.05)$, where $P \equiv psi/ft$

Warning: Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

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