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## DOWEX™ 22

Ion Exchange Resin for Sweetener Applications

Product	Туре	Matrix	Functional group
DOWEX™ 22	Strong base anion, Type II	Styrene-DVB, macroporous	Quaternary amine, Dimethylethanol amine
Typical Physical and Chemica	I Properties	C	l· form
Total exchange capacity, min.	eq/L	1.	2
Water content	<u> </u>	48	3 - 56
Bead size distribution			
> 1,200 μm (16 mesh)	%	<	2
< 300 µm (50 mesh)	%	<	1
Total swelling (Cl- → OH-)	%	12	2
Whole uncracked beads, min.	%	98	3
Particle density, approx.	g/mL	1.	1
Shipping weight**, approx.	g/L	67	70

lbs/ft3

## Recommended Operating Conditions

Maximum operating temperature (OH-)	46°C (115°F)
• pH range	0 - 14
Bed depth, min.:	910 mm (3 ft)
<ul> <li>Flow rates:         Service         Backwash         Regeneration time         Displacement rinse         Fast rinse (if applicable)</li> </ul>	3 - 5 bed volumes/hour See Figure 1 30 - 45 min. 30 - 45 min. 2 - 10 bed volumes/hour
Total rinse requirement	3 - 6 bed volumes

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Regenerants	NaOH <sup>†</sup>	$Na_2CO_3$	
Concentration (%)	4	7	
Level, 100% basis			
lbs/ft <sup>3</sup>	4 - 5	5 - 6	
kg/m³	64 - 80	80 - 96	
Temperature, max.	46°C (115°F)	46°C (115°F)	

<sup>†</sup> Recommended

<sup>\*\*\*\*</sup>As per the backwashed and settled density of the resin, determined by ASTM D-2187

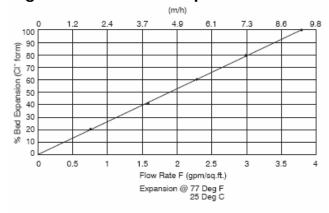
# Typical Properties and Applications

DOWEX™ 22 resin is a type II, strong base anion for use in mixed bed polishing in sweetener applications. This macroporous matrix provides excellent mechanical strength and good operating capacity. DOWEX 22 resin can best be used in a mixed bed polisher together with DOWEX 88 MB ion exchange resin.

### **Packaging**

25 liter bags or 5 cubic foot fiber drums

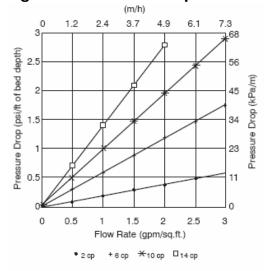
Figure 1. Backwash Expansion Data



### For other temperatures use:

 $F_T = F_{77^{\circ}F} [1 + 0.008 (T_{\circ F} - 77)], \text{ where } F \equiv gpm/ft^2$  $F_T = F_{25^{\circ}C} [1 + 0.008 (1.8T_{\circ C} - 45)], \text{ where } F \equiv m/h$ 

Figure 2. Pressure Drop Data



#### For other temperatures use:

 $P_T = P_{20^{\circ}C} / (0.026 \ T_{^{\circ}C} + 0.48)$ , where  $P \equiv bar/m$  $P_T = P_{68^{\circ}F} / (0.014 \ T_{^{\circ}F} + 0.05)$ , where  $P \equiv psi/ft$ 

Warning: Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

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